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Descargar libro gratis de 614,2 MB - Descargar Libros De Tiempo. Descarga presente del libro de gÃ³mez ireles y descarga gratis en formato epub pdf rar libro sobre la Applications of Spectroscopic Methods in Drug Design. The most important requirements of drug candidates in the pharmaceutical industry are safety, efficacy, and, last but not least, cost-effectiveness. Being a biomarker, drug metabolism is responsible for the safety of the drug molecules. In addition, pharmacokinetics and pharmacodynamics are the two main elements which influence drug efficacy. Therefore, the design of drugs, which are able to ensure the safety, efficacy, and cost-effectiveness of the drug, is a challenging task. These factors should be thoroughly investigated before drug development. Nowadays, spectroscopic methods are often involved in drug design. This paper provides an overview of the recent applications of spectroscopic methods in drug design. Starting with a short review of the elements which are important for drug development, we discuss (a) the determination of the physicochemical properties of the drug candidate, (b) the metabolism of drugs, (c) the characterization of drug molecules before taking them to the clinic, and (d) the intracellular localization of drugs. Inducing the Presence of an Intercalator for the Bulk Formation of Defect-Assisted Perovskites. Lead-based perovskites typically form in the presence of lead halides and/or intercalants. Herein, a double intercalation procedure is shown to induce the formation of defect-free lead-based perovskites. When a pillarized monolith intercalant is successively introduced to an etch-induced perovskite cavity of a mesoporous silica, the triple intercalation occurs at the vicinity of the cavity, while the single intercalation takes place inside the perovskite at the interior of the cavity. This procedure enables the bulk formation of defect-free lead-based perovskites, while the single intercalation can lead to the bulk formation of perovskites with higher ionic conductivity (up to 1500 S cm<sup>-1</sup>) and larger ionic separation ( $\Delta\mu > 3.00$  eV). The double intercalation is favorable for the formation of perovskites with large Brunauer-Emmett-Teller (BET) surface area.

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